

Unit 3 Notes Periodic Table Notes

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Unit 3 Notes Periodic Table

Unit 3 Notes: Periodic Table Notes John Newlands proposed an organization system based on increasing atomic mass in 1864. He noticed that both the chemical and physical properties repeated every 8 elements and called this the ____Law of Octaves _____. In 1869 both Lothar Meyer and Dmitri Mendeleev showed a connection

Unit 3 Notes: Periodic Table Notes

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Unit 3 Notes: Periodic Table Notes

Unit 3: Periodic Table - CHEMISTRY WITH MS.KING. CHEMISTRY WITH MS.KING. Periodic Trends Powerpoint. Periodic Trends Notes. The chart below shows how many electrons are on the last ring of each element. Elements are grouped in this way. Each period (row) represents how many rings an atom would have.

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The Atomic Mass Unit (AMU) is defined as 1/12th the mass of a carbon-12 atom, or 1.66*10-24 g. Reading the periodic table. Groups (families) of elements (vertical columns on the periodic table) Elements that are part of the same group behave in a chemically similar way. The chemical symbol is an abbreviation of the element’s name. Periodic Law

Unit 3 Notes - The Periodic Table

Periodic Law = elements in periodic table are PERIODIC functions of their ATOMIC NUMBER 1) As you move down a GROUP or COLUMN, you add 1 PRINCIPAL ENERGY LEVEL or ELECTRON SHELL 2) As you move across a ROW or PERIOD, you add 1 PROTON to the nucleus, and 1 ELECTRON to the VALENCE SHELL 3) The MAXIMUM number of VALENCE ELECTRONS any

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EXAM 4th Jun 09 @ 9am [CHEMISTRY UNIT 3 REVISION NOTES] Page 1 of 14 14 Brief History of Periodic Table • Until quite recently, there were only two ways of categorising elements: o Their physical and chemical properties o Their relative atomic mass • In 1864, John Newlands tried to arrange the elements with his law of octaves. • His work was ignored as there was inconsistency as he ...

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Unit 3 - The Periodic Table VOCABULARY: Ionization energy Electronegativity Atomic Radius Ionic Radius Chemical Reactivity Metallic Character Nonmetallic character Metals Metalloids Nonmetals Alkali metals Alkaline Earth metals Halogens Noble Gases Transition metals Periodic Periodic Law Periods Groups Octet States of matter Solids Liquids Gases

Mr. Dolgos Regents Chemistry NOTE PACKET Unit 3: Periodic ...

Groups 3 through 8 are called the main block elements. The metals in this group are aluminum, gallium, indium, tin, thallium, lead, bismuth, & polonium. The metalloids in this group are boron, silicon, germanium, arsenic, antimony & tellurium. The nonmetals in this group are hydrogen, oxygen, nitrogen,

Unit 3.2: The Periodic Table and Periodic Trends Notes

Classification of Elements and Periodicity in Properties Class 11 Notes Chemistry Chapter 3 • Genesis of Periodic Classification Dobereiner’s Triads In 1829, Dobereiner arranged certain elements with similar properties in groups of three in such a way that the atomic mass of the middle element was nearly the same as the average atomic masses of [...]

Classification of Elements and Periodicity in Properties ...

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Unit 3 - The Periodic Table VOCABULARY: Ionization energy Electronegativity Atomic Radius Ionic Radius Chemical Reactivity Metallic Character Nonmetallic character Metals Metalloids Nonmetals Alkali metals Alkaline Earth metals Halogens Noble Gases Transition metals Periodic Periodic Law Periods Groups Octet States of matter Solids Liquids Gases

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PERIODIC TABLE: Dmitri Mendeleev –mid 1800’s-proposed a table for 70 elements based on increasing mass and similar properties Henry Moseley –1913-determined the atomic number of elements and arranged the table in order of increasing atomic number

CHAPTER 6 NOTES: The Periodic Table

PERIODIC TABLE UNIT 3. Periodic Table PPT. Periodic Table Notes Outline. Periodic Table ActivityButton. Periodic Table Webquest. online activity ...

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9th Chemistry Unit#3 Periodic Table ... - Top Study Notes

History of the Periodic Table & Organization of the Periodic Table Notes ... To access the Module 3 Life Sized Periodic Table Project Information: - Scroll to the top & hover over "Chemistry" until a drop down list appears - Move your mouse to hover over Unit 3 and you will see another pop out menu for the project pop out.

Unit 3 - The Atom & the Periodic Table - Mrs. Ashley's ...

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Unit 3 - Periodic Table - Mrs. Watson's Science Class

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