

Pogil The Activity Series Answer Key

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Pogil The Activity Series Answer

Model 2 — The Activity Series 10. The activity series shown in Model 2 to the right is a ranking, similar to the one you did in Question 7, of several metals. a. Are the more active metals at the top or at the bottom? Explain. ecu.' (b. Label Model 2 with "most active" and "least active" in the appropriate locations. 11. Use the activity ...

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CuCl was named copper (I) chloride because copper can have two possible charges (a +1 or a +2) so you need to use the Roman numeral (I) to show that this compound was made with copper with a +1 charge. Name the following compounds. ZnCl2 _____, CoN _____, PbS2 _____.

Chemistry POGIL Activity «Activity_»

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Pogil Activity Worksheets - Teacher Worksheets

Pogil The Activity Series Answer Key The reactivity of the metals is exactly opposite the reactivity of the corresponding cation. While solid magnesium was the most reactive metal, its ion, Mg²⁺, was the least reactive. - 2+Mg + 2e -> Mg²⁺ -Zn + 2e -> Zn -2+Cu +

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